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#### Key indicators

Single-crystal X-ray study  
T = 293 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$   
R factor = 0.037  
wR factor = 0.104  
Data-to-parameter ratio = 14.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

# 1'-Methyl-5-diphenyl-4'-p-tolyl-5,6,7,8,9,10-hexahydro-1,3-cyclooctapyrimidino[2,3-b]thiazole-2-spiro-3'-pyrrolidine-2'-spiro-3''-1H-indole-2'',3(2H,3''H)-dione

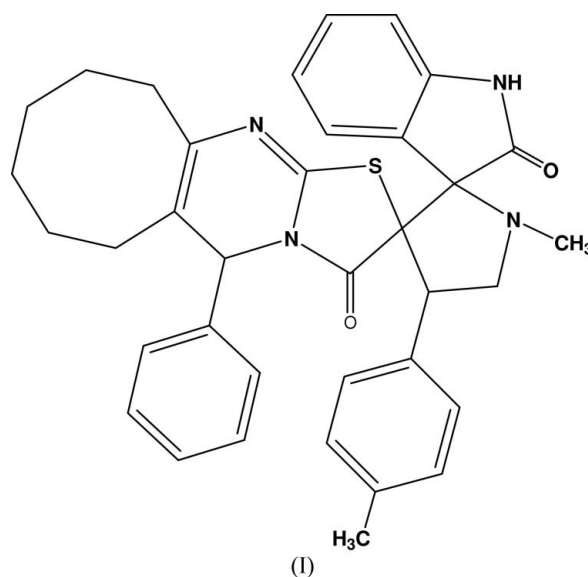
The pyrrolidine ring in the title compound,  $\text{C}_{36}\text{H}_{36}\text{N}_4\text{O}_2\text{S}$ , adopts an envelope conformation. The thiazolidine ring adopts a twisted conformation, with a pseudo-twofold axis passing through the N atom and the opposite S—C bond. The crystal packing is stabilized by  $\text{N}-\text{H}\cdots\text{N}$  intermolecular interactions.

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#### Comment

Pyrrolidine, a basic intermediate used in a wide range of applications in organic synthesis, has gained much attention in the pharmacological industry for its medicinal value. Derivatives of pyrrolidine are found to have anticonvulsant (Obniska *et al.*, 2002), antimicrobial and antifungal activity against various pathogens, except *Bacillus subtilis* (Amal Raj *et al.*, 2003). Oxindole derivatives help to treat and prevent diabetic complications arising from elevated levels of sorbitol and act as aldose reductase inhibitors (Rajeswaran *et al.*, 1999). Thiazolidines and their derivatives have diverse biological importance with properties such as local anaesthetic, anti-seizure, antitubercular, antibacterial, anti-amoebic, anti-diabetic and anti-inflammatory (Chaurasia, 1971; Zimenkovsky *et al.*, 1999), and anti-oxidant (Rachinskii, 1964). The spiro ring system is a frequently encountered structural motif in many pharmacologically relevant alkaloids.



As spiro-pyrrolidine compounds are of great medicinal importance, we have undertaken the three-dimensional structure determination of the title compound, (I), by X-ray diffraction (Fig. 1).

The molecular geometry of (I) is comparable with that of a similar structure reported earlier (Gayathri *et al.*, 2005). The sums of the bond angles around N1 (337.0°) and N3 (359.4°) indicate  $sp^3$  and  $sp^2$  hybridization, respectively.

The methyl atom C5 lies 0.542 (2) Å below the plane of atoms C1–C4 and atom C36 lies 0.094 (3) Å above the plane of the benzene ring C30–C35. The five- (C1/C6/N2/C7/C12) and six- (C7–C12) membered rings in the indane group are planar, with a dihedral angle of 5.8 (1)° between these rings. Atom O1 deviates by 0.073 (1) Å from the plane of the five-membered ring in the indane group. The six-membered ring N3/C14/C21/C28/N4/C29 is slightly non-planar, with atom C14 deviating by 0.230 (2) Å from the plane of the other atoms, because of the phenyl (C15–C20) substituent at atom C14. The dihedral angle between the two benzene rings (C15–C20 and C30–C35) is 19.6 (1)°. The pyrrolidine ring adopts an envelope conformation, with atom N1 deviating by 0.588 (1) Å from the mean plane of atoms C1–C4, and the thiazolidine ring adopts a twisted conformation with a pseudo-twofold axis passing through N3 and the S1–C2 bond.

The puckering parameters (Cremer & Pople, 1975) and the smallest displacement asymmetry parameter (Nardelli, 1983) for the pyrrolidine ring are  $q_2 = 0.409$  (2) Å,  $\varphi = 355.3$  (3)° and  $\Delta_s(N_1) = 5.0$  (2), and for the thiazolidine ring are  $q_2 = 0.084$  (1) Å,  $\varphi = 18.5$  (11)° and  $\Delta_2(N_3) = 0.9$  (2).

The molecule of (I) is stabilized by weak C–H...O intramolecular interactions. The crystal packing is stabilized by N–H...N intermolecular interactions, generating a centrosymmetric dimer of  $R_2^2(16)$  motif (Bernstein *et al.*, 1995) (Table 1 and Fig. 2).

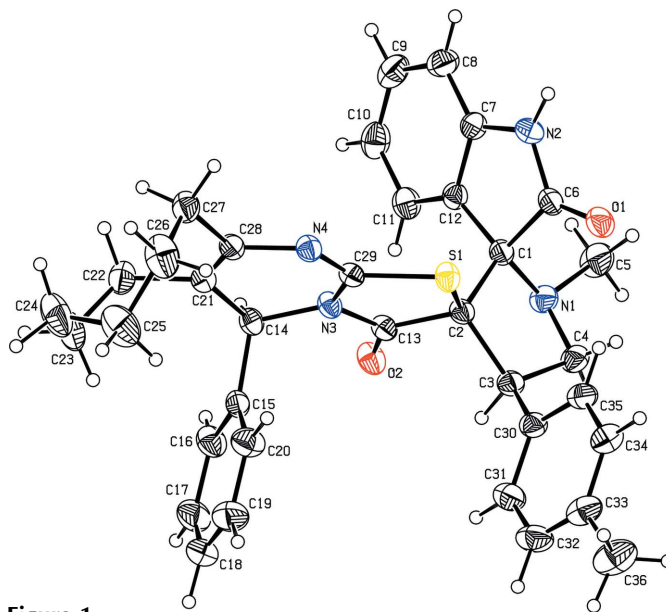
These results, except for the different conformation of the thiazolidine ring, are very similar to those for the closely related compound reported in the preceding paper (Gayathri *et al.*, 2006).

## Experimental

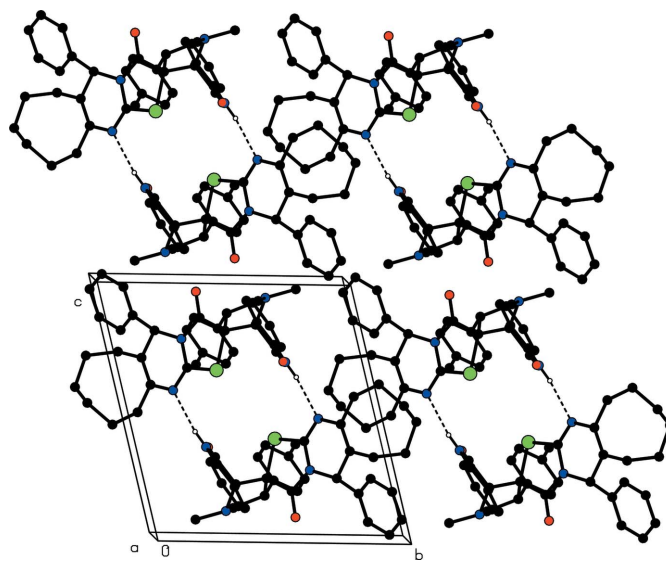
A mixture of isatin (1.2 mmol), sarcosine (1.2 mmol) and 5-phenyl-2-(*p*-methyl)phenylmethylene-5,6,7,8,9,10,11-heptahydrocycloocta-[*d*]thiazolo[3,2-*a*]pyrimidin-3(2*H*)-one (1 mmol) in methanol-dioxane (1:1, 20 ml) was refluxed until the disappearance of the starting materials (4 h) as shown by thin-layer chromatography analysis. The reaction mixture was then concentrated *in vacuo* and extracted with water (50 ml) and dichloromethane (50 ml). The organic layer was washed with brine, dried with anhydrous sodium sulfate and concentrated *in vacuo*. The residue was purified by column chromatography (silica gel, 100–200 mesh) with a hexane-ethyl acetate (8:2) mixture to give the title compound, which was recrystallized from methanol by slow evaporation.

### Crystal data

$C_{36}H_{36}N_4O_2S$	$V = 1536.88$ (19) Å <sup>3</sup>
$M_r = 588.75$	$Z = 2$
Triclinic, $P\bar{1}$	$D_x = 1.272$ Mg m <sup>-3</sup>
$a = 11.3531$ (8) Å	Mo $K\alpha$ radiation
$b = 12.1973$ (9) Å	$\mu = 0.14$ mm <sup>-1</sup>
$c = 13.2117$ (9) Å	$T = 293$ (2) K
$\alpha = 94.181$ (1)°	Block, colourless
$\beta = 111.683$ (1)°	$0.25 \times 0.22 \times 0.20$ mm
$\gamma = 111.377$ (1)°	



**Figure 1**  
The molecular structure of the title compound, showing 30% probability displacement ellipsoids.



**Figure 2**  
The packing of (I), viewed approximately down the *a* axis. Hydrogen bonds are shown as dashed lines. H atoms not involved in hydrogen bonding have been omitted.

### Data collection

Bruker SMART APEX CCD area-detector diffractometer	5422 independent reflections
$\omega$ scans	4781 reflections with $I > 2\sigma(I)$
Absorption correction: none	$R_{int} = 0.019$
14952 measured reflections	$\theta_{max} = 25.0^\circ$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0549P)^2 + 0.3799P]$
$R[F^2 > 2\sigma(F^2)] = 0.037$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.104$	$(\Delta/\sigma)_{max} = 0.001$
$S = 1.03$	$\Delta\rho_{max} = 0.27$ e Å <sup>-3</sup>
5422 reflections	$\Delta\rho_{min} = -0.17$ e Å <sup>-3</sup>
388 parameters	
H-atom parameters constrained	

**Table 1**  
Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N2-H2\cdots N4^i$	0.86	2.13	2.975 (2)	170
$C3-H3\cdots O2$	0.98	2.43	2.932 (2)	111
$C4-H4A\cdots O1$	0.97	2.50	3.069 (2)	117
$C11-H11\cdots O2$	0.93	2.49	3.097 (2)	123
$C35-H35\cdots O1$	0.93	2.58	3.426 (3)	152

Symmetry code: (i)  $-x + 2, -y + 1, -z + 1$ .

All H atoms were treated using a riding model, with C–H = 0.93 for aromatic H, 0.98 for methine H, 0.97 for methylene H and 0.96 Å for methyl H, and N–H = 0.86 Å, and with  $U_{iso}(H) = 1.2U_{eq}(C,N)$ , or  $1.5U_{eq}(C)$  for methyl groups.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINTE* (Bruker, 2001); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1995).

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